

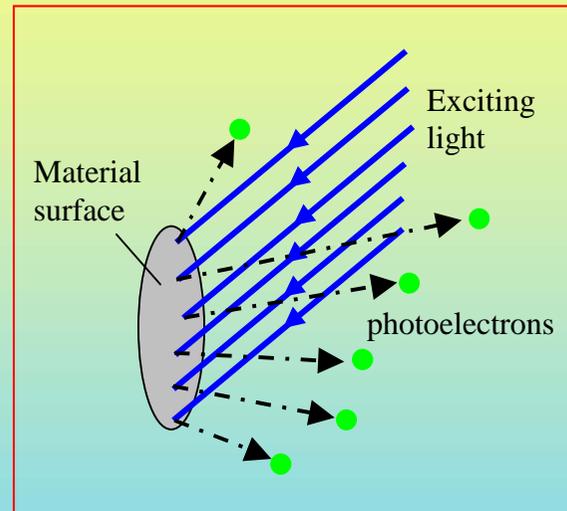
Photonics



- Photonics is the branch of the study of light where photons are important. Notably:
 - ▶ the detection of light
 - ▶ the generation of light
 - ▶ quantum optics
- The name is commercially popular
- This section concentrates on the detection of light and, in particular, methods involving the photoelectric effect

Photoelectric effect

- The release of electrons upon bombardment with light
- Einstein
 - ▶ Nobel prize
- 2 key properties:
 - ▶ below a certain frequency no electrons are ever produced
 - ▶ the kinetic energy of the electrons increases with light frequency



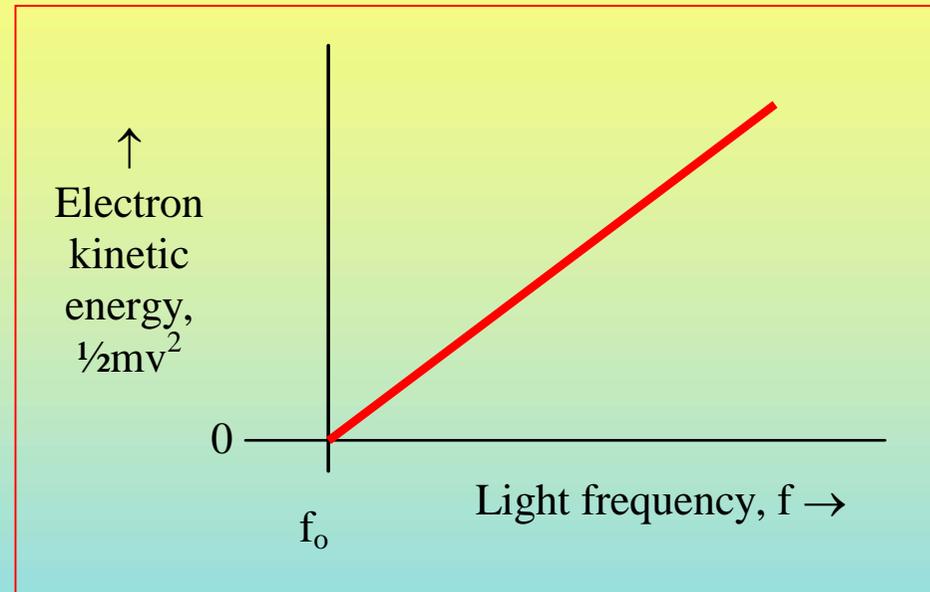
Albert Einstein 1879 - 1955

Quantitative results

- Einstein postulated that light consisted of nothing but quanta – later called photons
- Planck's formula

$$Energy = hf$$

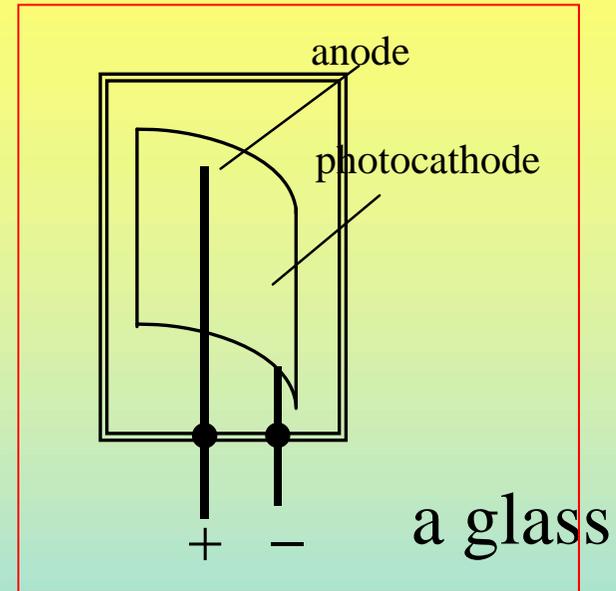
- Einstein's interpretation: energy conservation
- Φ is the work function of the material



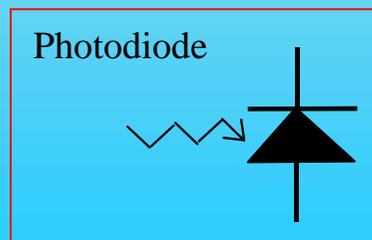
$$\frac{1}{2}mv^2 = hf - \Phi$$

Photocells

- Photocells deliver a photocurrent when put into a circuit
 - ▶ the photocell is housed in vacuum tube



- Photodiodes are solid-state devices generating electrons in response to light

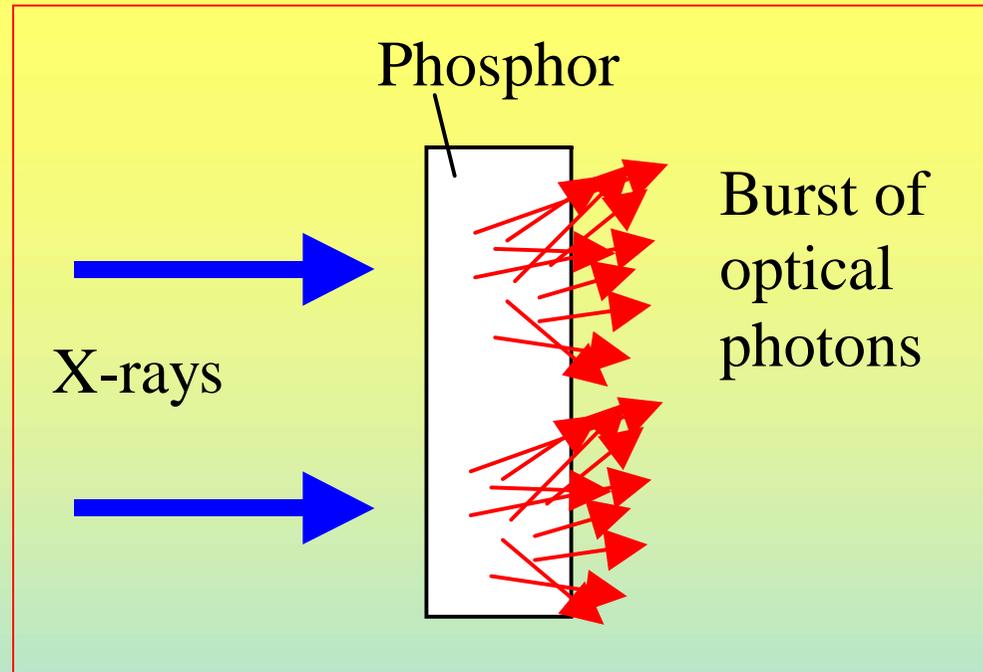


The photomultiplier

- Photomultipliers are used to detect weak light sources
- Photomultipliers are used to detect X-rays and γ -rays
 - ▶ the detection is a 2-stage process:
 - the radiation produces light via phosphors
 - the light is then detected by a photomultiplier

Phosphors

- X-ray photons may have 10,000 times the energy of an optical photon



- One X-ray photon creates several thousand optical photons
 - ▶ the time taken is very short $\sim 10^{-7} - 10^{-10}$ s
- Suitable materials are some plastics and selected doped crystals

Working of the photomultiplier

- The photomultiplier amplifies the initial photoelectron pulse by a succession of dynodes each $\sim 100\text{V}$ more positive giving a total amplification $>10^6$

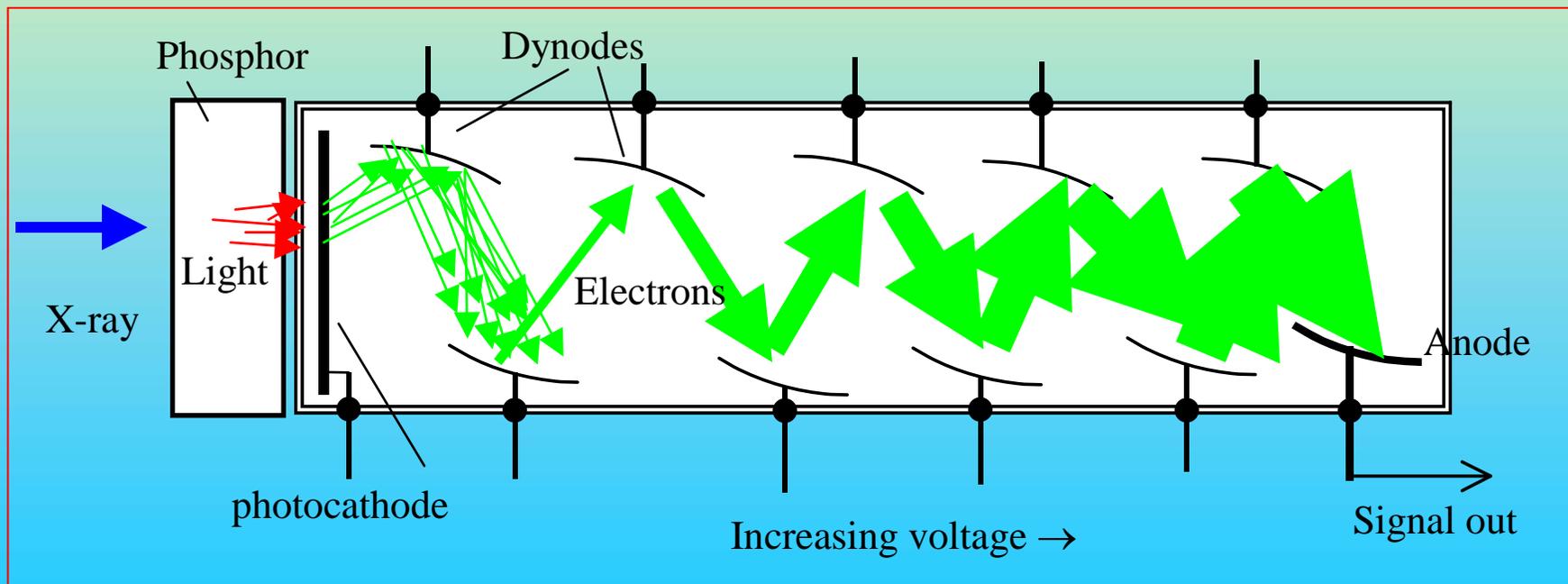
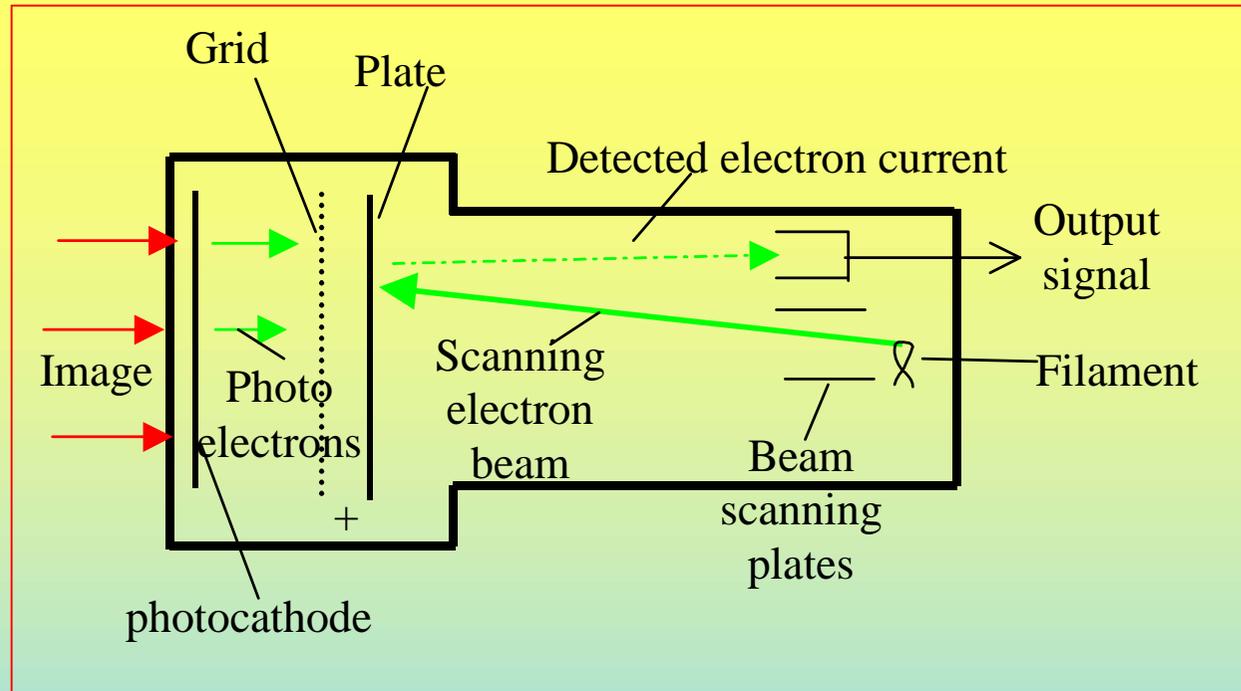


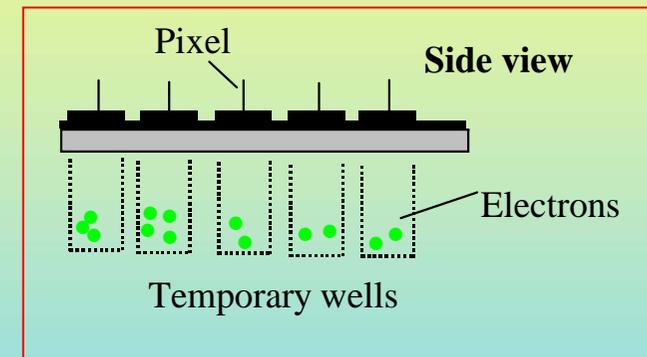
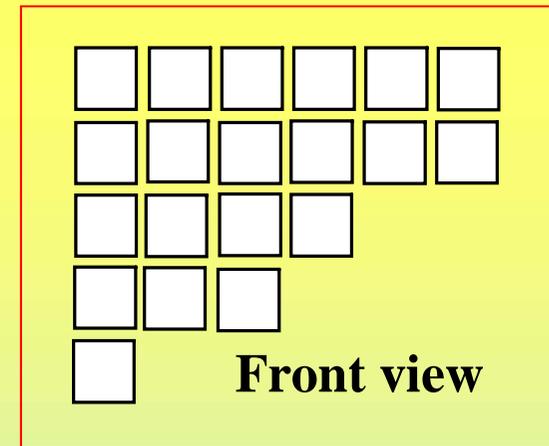
Image intensifiers



- The image orthicon is based on photoelectron emission
- The photoelectrons create a +ve replica of the image on the internal plate
- This is scanned by an electron beam and the scattered electrons are measured

The CCD detector

- **Charge Coupled Device**
- Expose the detector to light, generating the photoelectrons
- Store the photoelectrons in electrical ‘wells’ directly beneath their place of generation
- Transfer the stored charges to the edge of the device
- Read-out the charge and generate a suitable digital number to represent the light irradiance

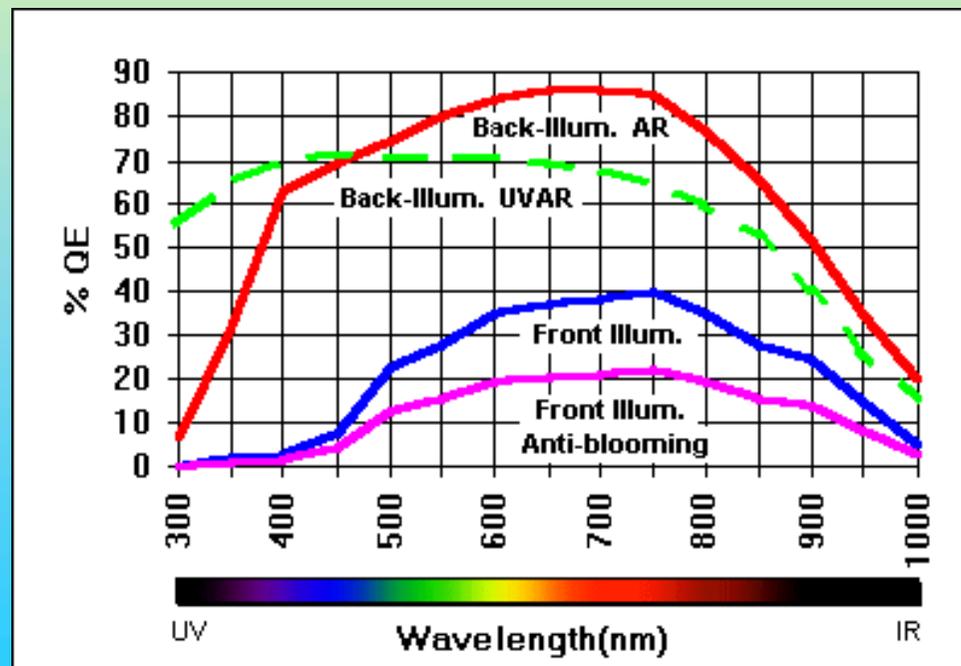


Advantages of the CCD

- They are intrinsically digital
- They're highly linear: twice the irradiance produces twice the output voltage
- They have a large dynamic range (the maximum to minimum levels that can be detected)
- They can integrate the light for many seconds, allowing the imaging of sources too weak to be seen by eye
- They have a low 'noise', or dark signal, made even lower by cooling. A few electrons per pixel is achievable

The quantum efficiency

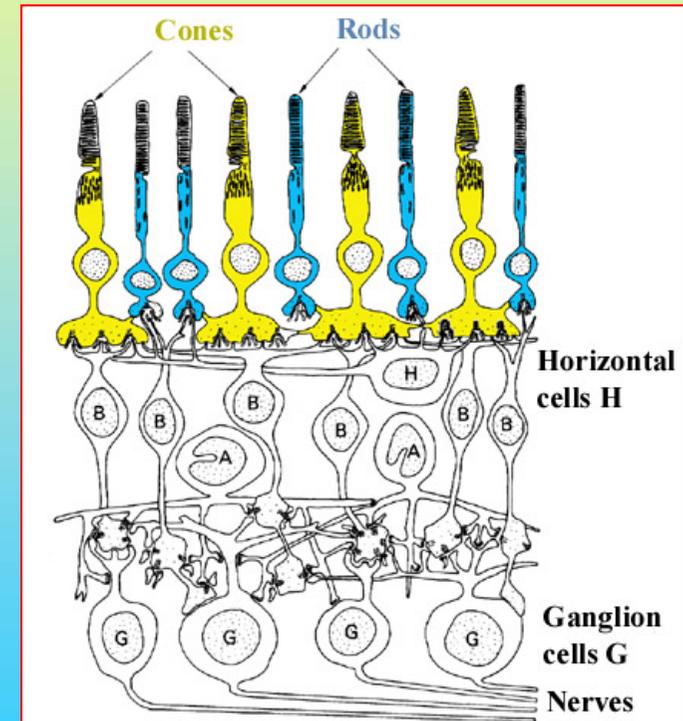
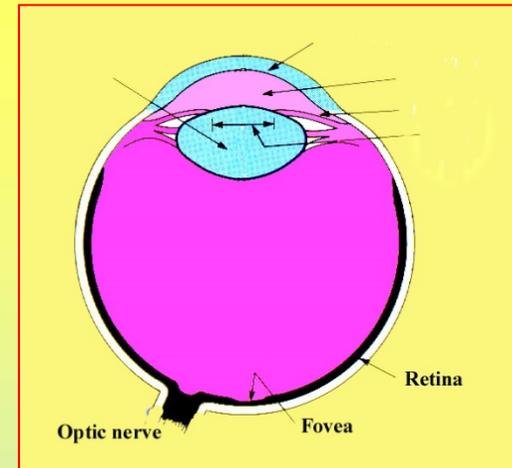
- The quantum efficiency is the success at creating a signal for incident quanta. It is very high indeed for CCD detectors



Courtesy:
apogee-ccd.com

Detecting light by eye

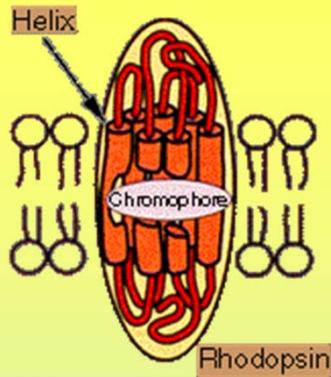
- The eye takes in light and produces electrical signals
- The crucial molecule is a protein called rhodopsin
- Within rhodopsin is a chromophore that absorbs light, called retinal



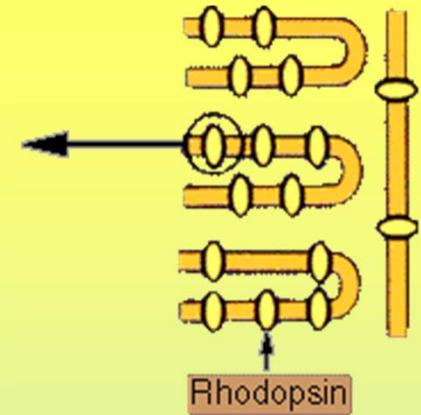
After Dowling & Boycott

Light

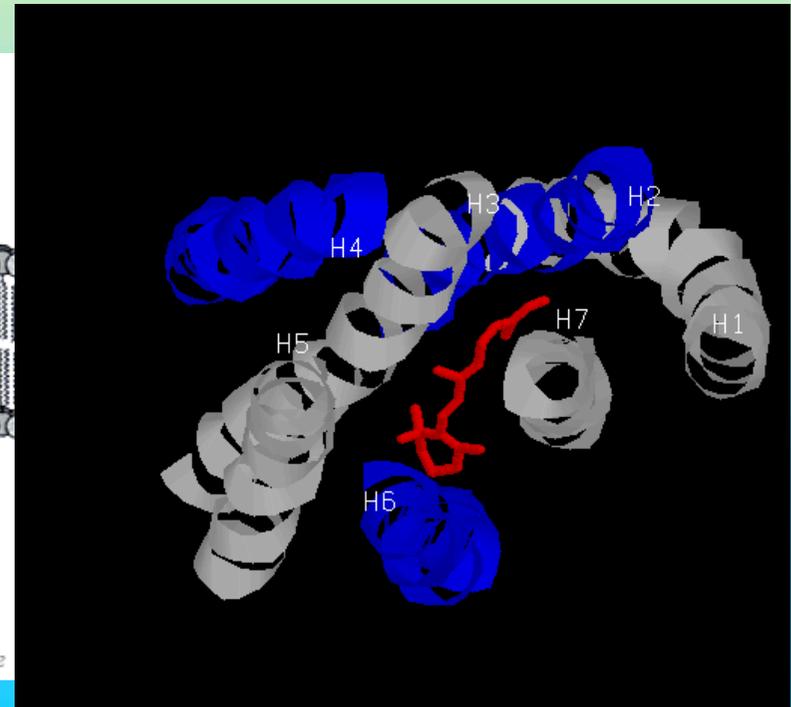
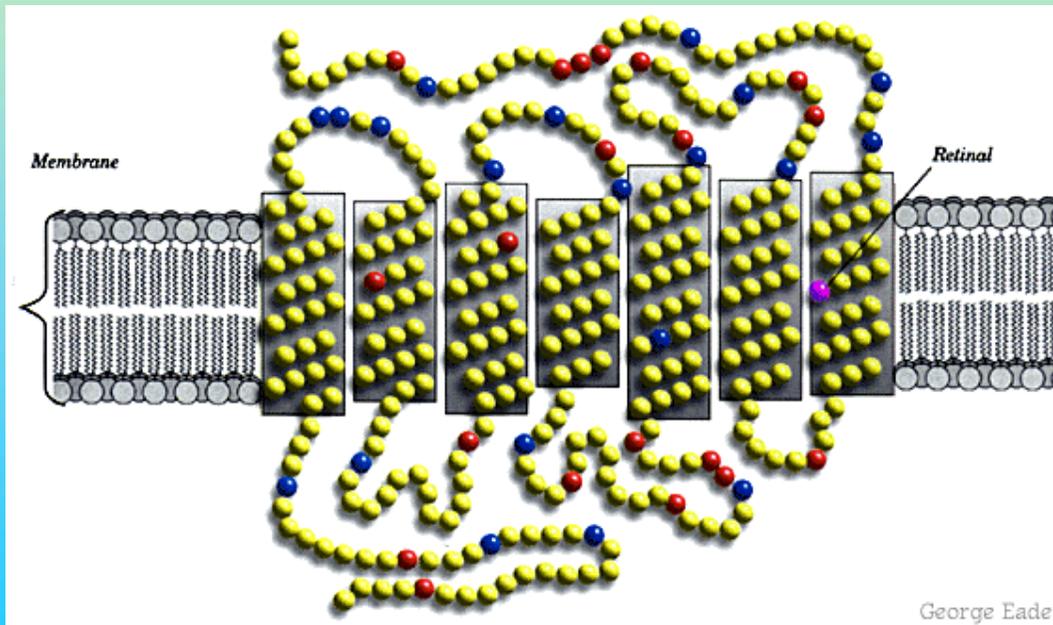
Six colored arrows (red, blue, green, yellow, orange, red) pointing upwards, representing light entering the eye.



Rhodopsin in rods

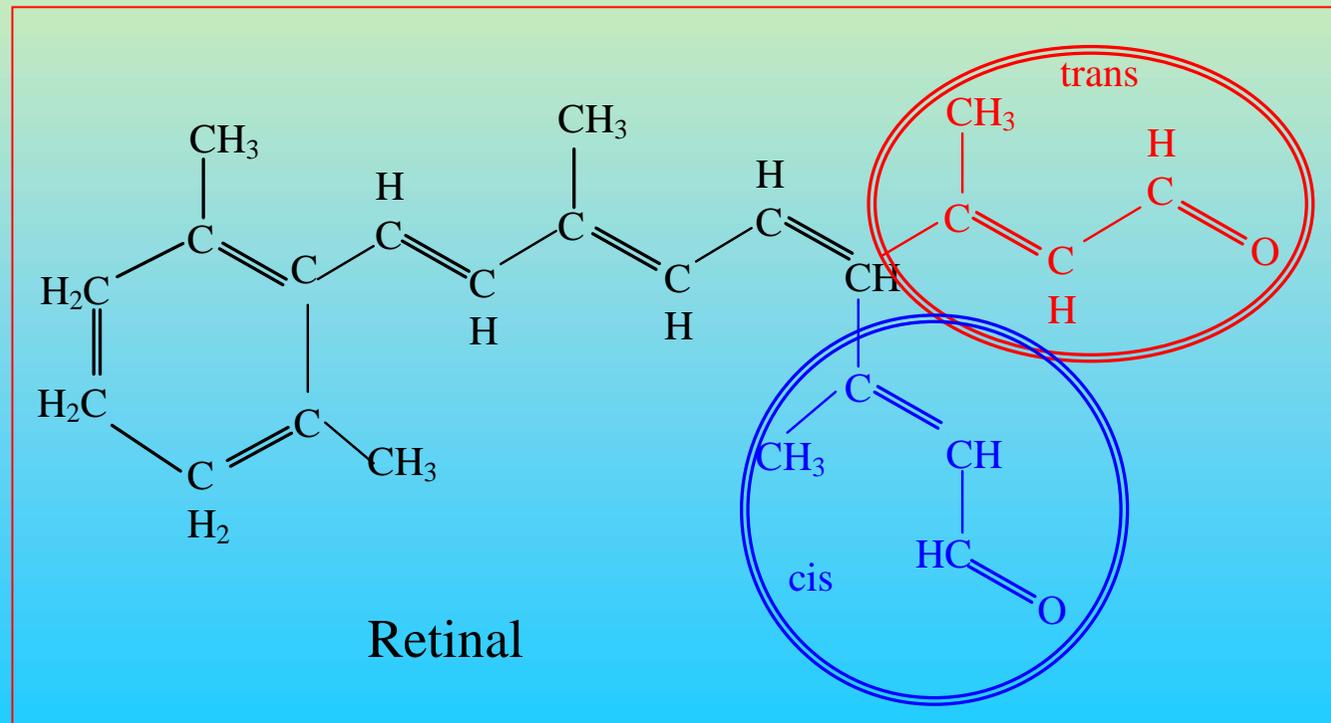


- Rhodopsin winds up and down a rod
- Notice the retinal



The working of retinal

- Upon absorbing light, retinal changes from the *cis* to the *trans* configuration, releasing an electron
- This is the start of a cascade process, leading to generation of an optic nerve signal



The end